

# **Pre-Lithiation of High-Capacity Battery Electrodes**

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Stanford University

June 13, 2019

Project ID  
bat272



# Overview

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## Timeline

- Start: August 1, 2017
- End: October 31, 2020
- Percent complete: 60%

## Budget

- Total project funding  
\$900k from DOE
- Funding for FY18  
\$300k
- Funding for FY19  
\$300k

## Barriers

### Barriers of prelithiation

- Low Coulombic efficiency
- Low capacity
- High chemical reactivity

**Targets:** high-efficiency and high-energy batteries

## Partners

- Collaboration
  - BATT program PI's
  - SLAC: In-situ X-ray
  - Amprius Inc.
  - Stanford: Zhenan Bao



# Project Objective and Relevance

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## Objective

- Design and synthesize lithiated silicon to prelithiate various anode materials.
- Increase first-cycle Coulombic efficiency via anode prelithiation.
- Increase first-cycle Coulombic efficiency via cathode prelithiation.
- Increase the stability of prelithiation reagents in both dry air and ambient air conditions.
- Design and fabricate fully lithiated anode materials to pair with high capacity lithium-free cathodes for next generation high energy density batteries.

# Milestones for FY18 and 19

Month/year	Milestones
1/2018	Develop Li-containing anodes with excellent electrochemical and environmental stability (completed)
4/2018	Develop Li-containing anodes with excellent rate capability (completed)
7/2018	Develop Li-containing anodes with excellent cycling stability for Li-S full cell (completed)
10/2018	Fabricate free-standing $\text{Li}_x\text{Sn}$ /graphene or $\text{Li}_x\text{Al}$ /graphene foil as an alternative to lithium metal anodes (completed)
1/2019	Demonstrate anode prelithiation reagent with specific capacity $>900 \text{ mAh/g}$ with stability in ambient air (completed)
4/2019	Demonstrate anode prelithiation process to be compatible with a variety of solvent processing (on track)

# Approach/Strategy

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## **Prelithiation reagents design and synthesis**

- 1) Compensate 1<sup>st</sup> cycle anode capacity loss with anode prelithiation reagents ( $\text{Li}_x\text{Si}$  nanoparticles)
- 2) Compensate the anode capacity loss with cathode prelithiation reagents ( $\text{Li}_2\text{O}$ /metal nanocomposite)
- 3) Achieve improved stability of anode prelithiation reagents in the dry and ambient air condition by exploring inorganic and organic coatings, such as  $\text{Li}_2\text{O}$ ,  $\text{LiF}$  and artificial SEI-coating.
- 4) Increase the stability of cathode prelithiation reagents by exploring different composites

## **Structure and property characterization**

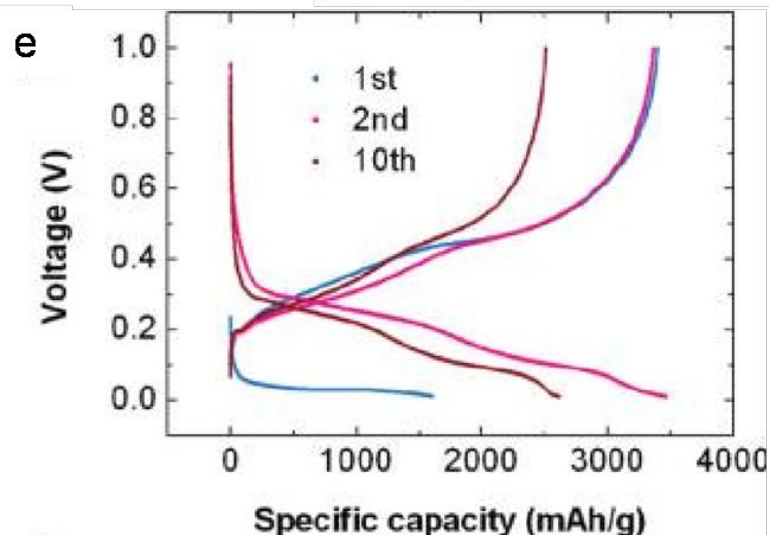
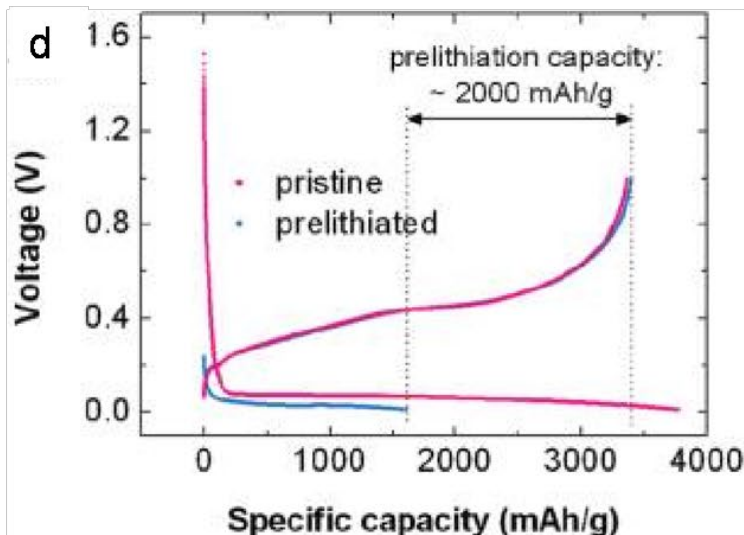
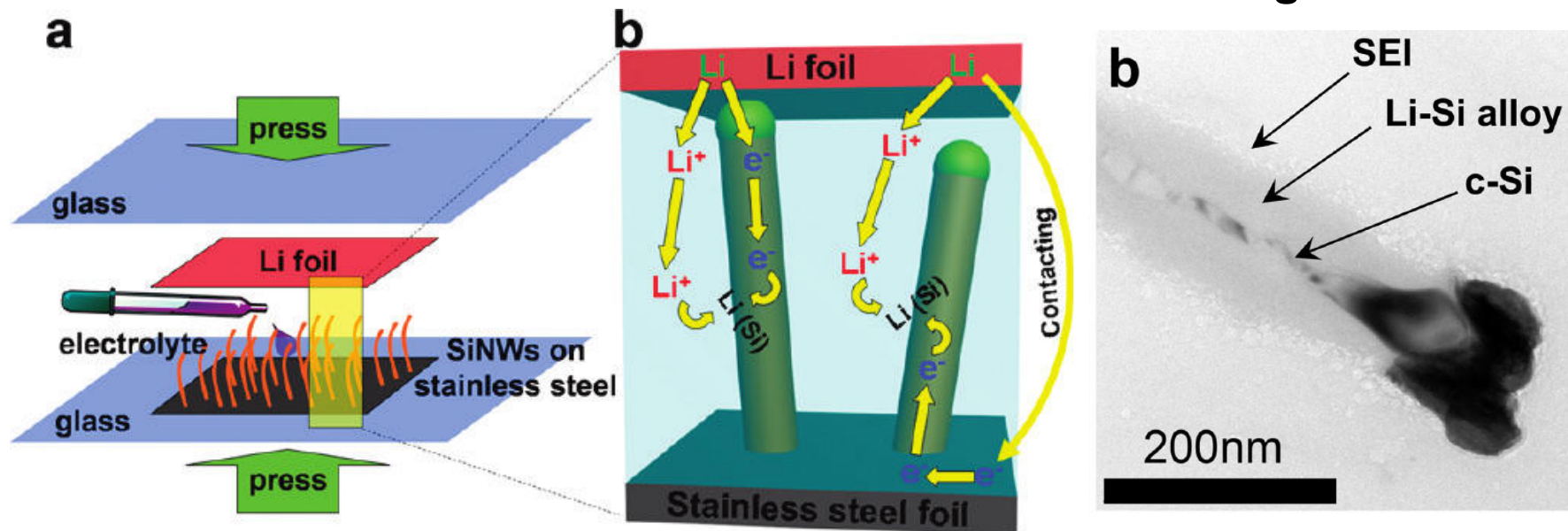
- 1) Ex-situ transmission electron microscopy
- 2) Ex-situ scanning electron microscopy
- 3) Ex-situ X-ray photoelectron spectroscopy
- 4) In operando X-ray diffraction and transmission X-ray microscopy

## **Electrochemical testing**

- 1) Coin cells and pouch cells
- 2) A set of electrochemical techniques

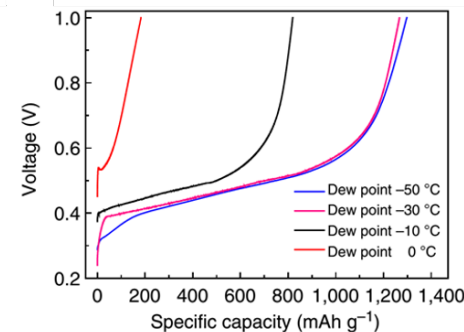
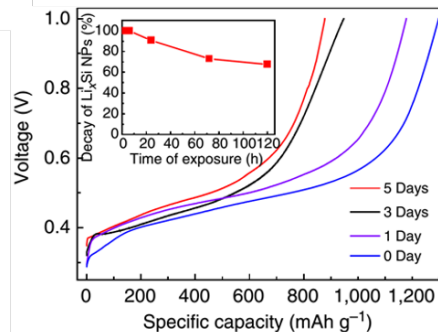
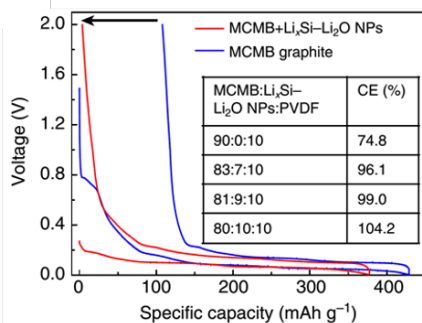
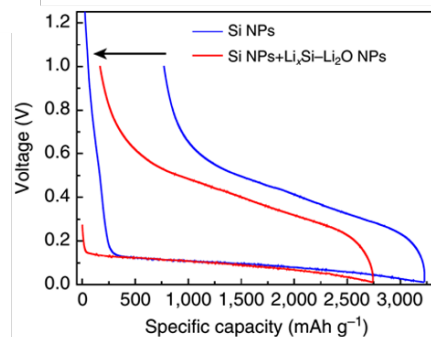
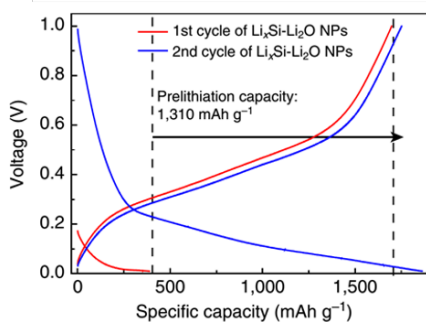
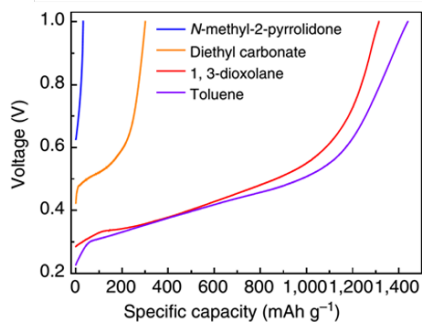
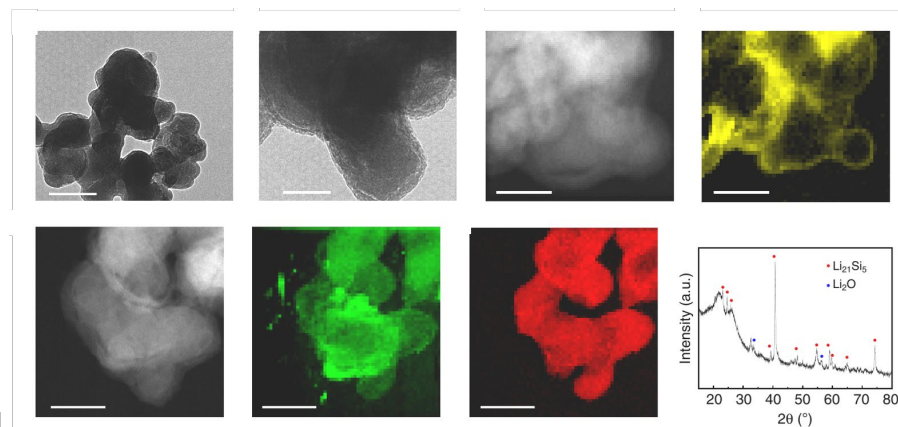
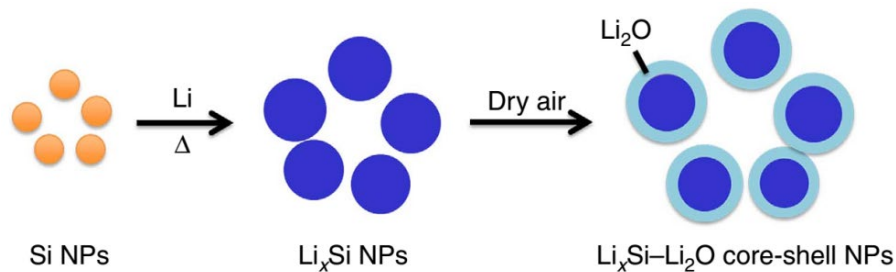
# Previous Accomplishments on Prelithiation

## Prelithiation of Si nanowires via electrochemical shorting mechanism



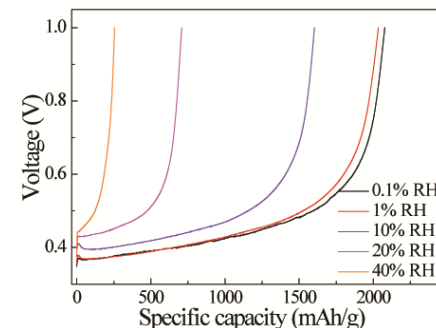
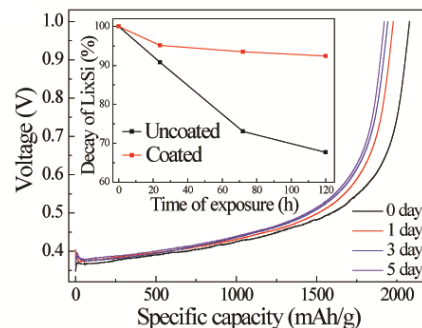
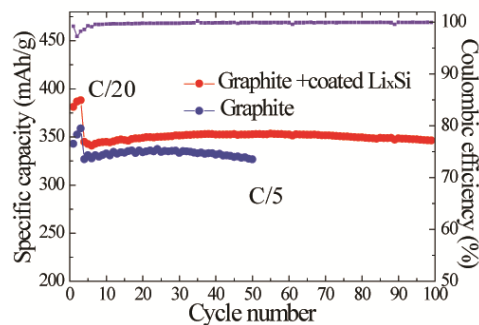
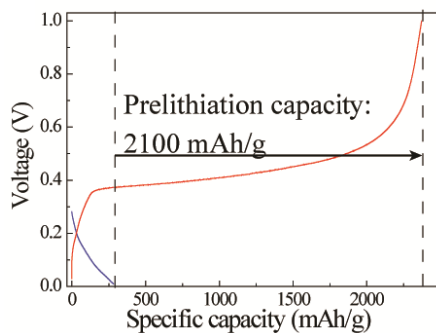
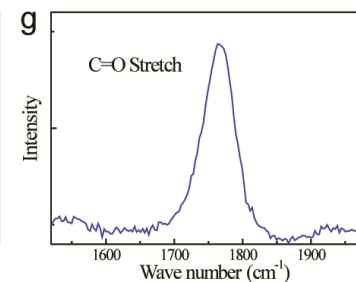
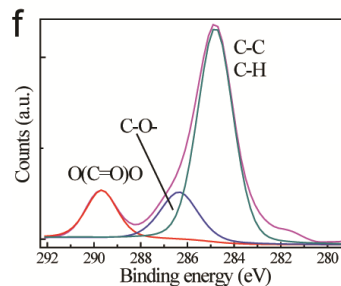
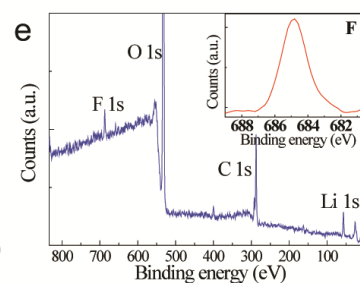
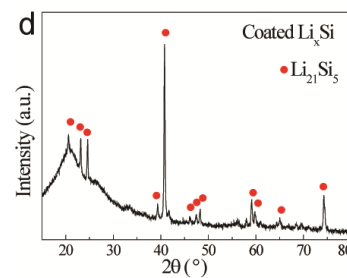
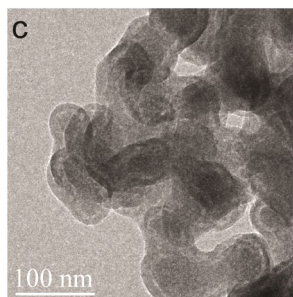
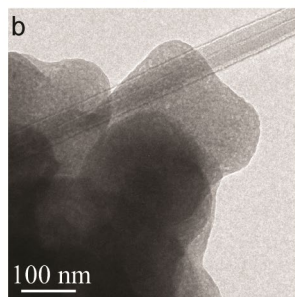
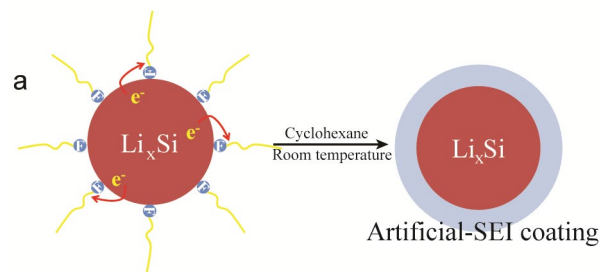
# Accomplishment

## Dry-air-stable $\text{Li}_x\text{Si-Li}_2\text{O}$ core-shell nanoparticles for anode prelithiation



Cui group, *Nat. Commun.* 5, 5088 (2014)

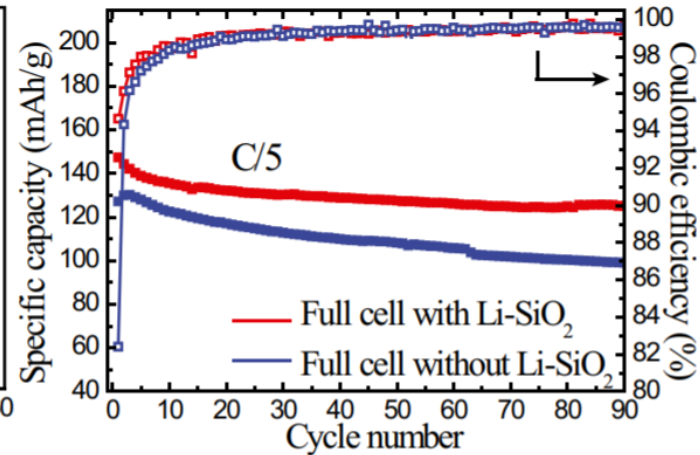
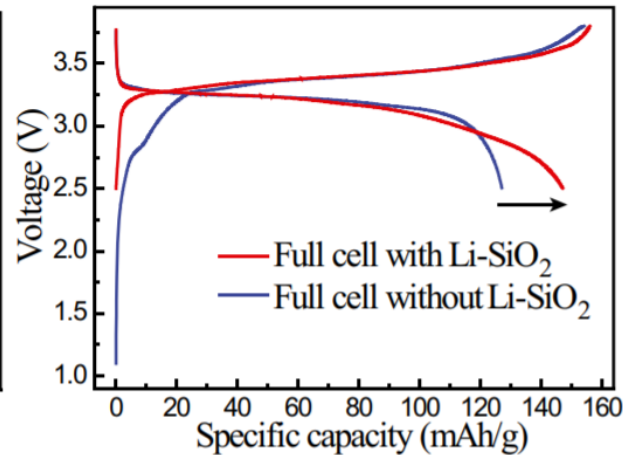
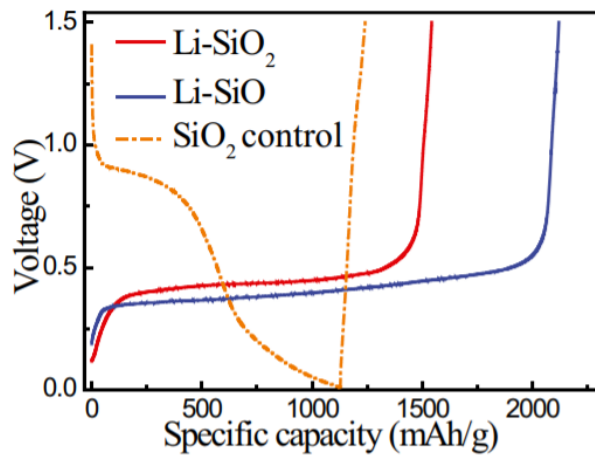
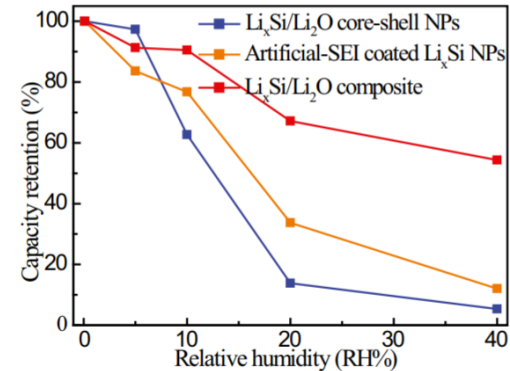
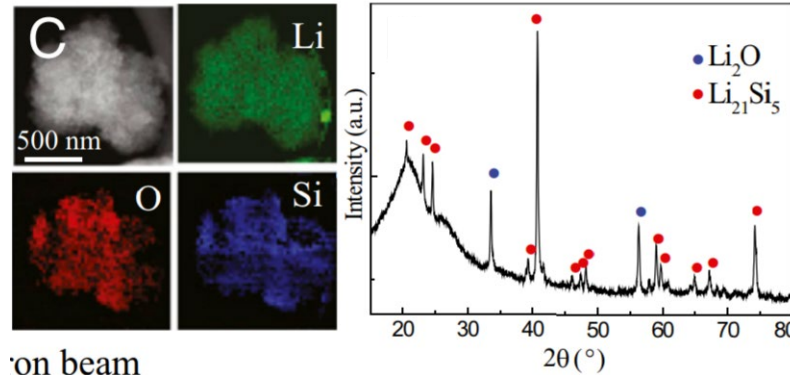
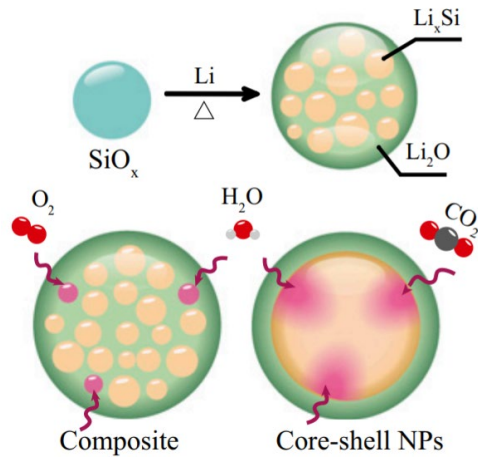
# Artificial-SEI protected $\text{Li}_x\text{Si}$ nanoparticles for anode prelithiation





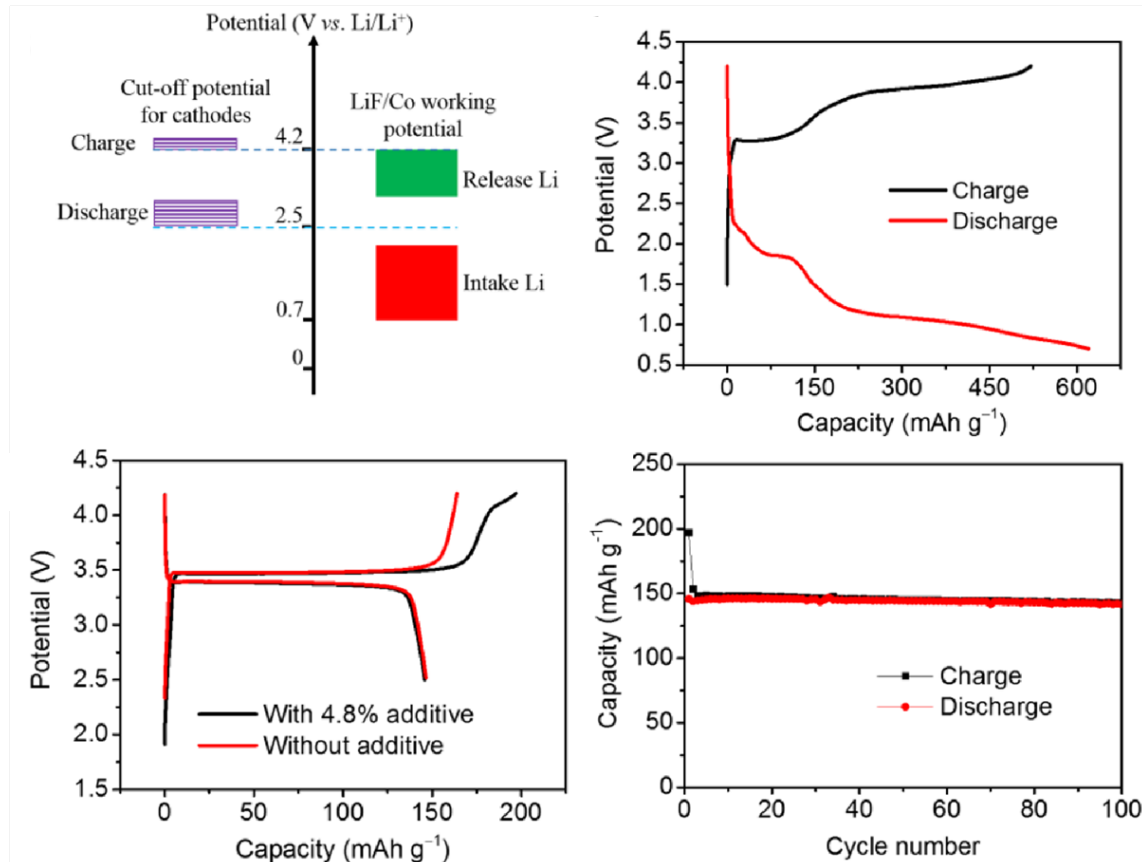
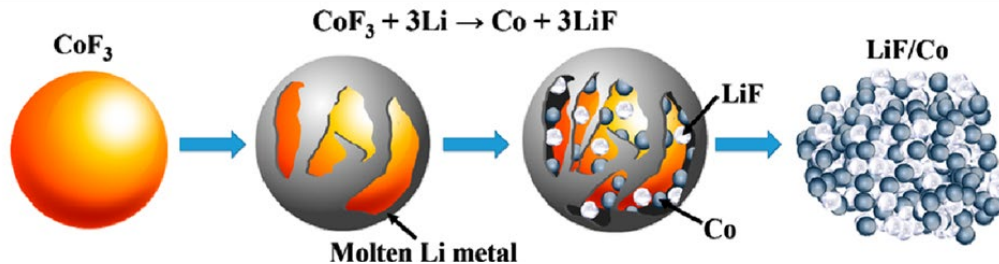
# Accomplishment

## Metallurgically lithiated SiO<sub>x</sub> for anode prelithiation



# Accomplishment

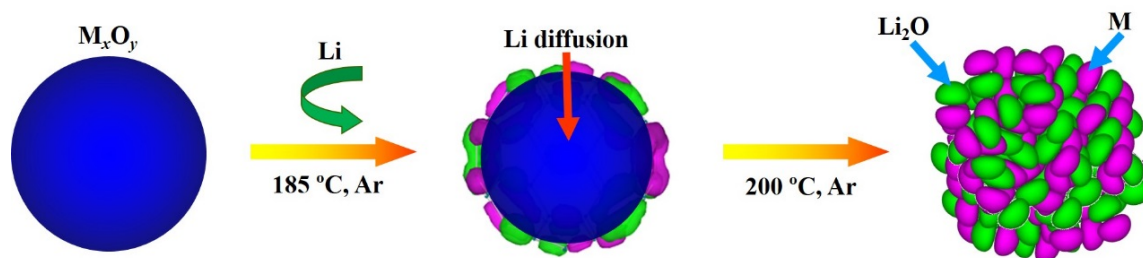
## LiF/metal nanocomposite for cathode prelithiation



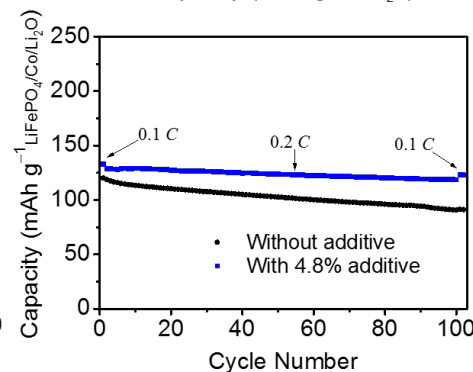
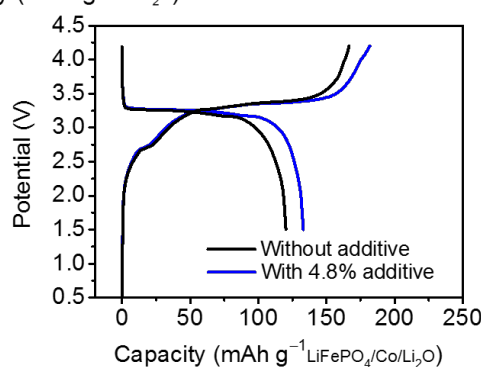
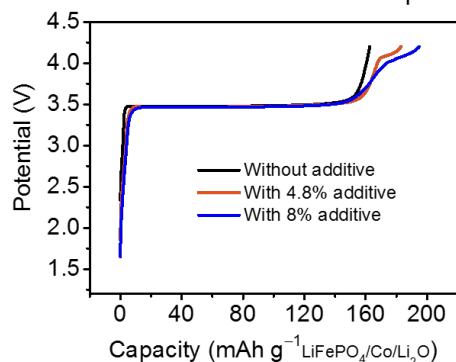
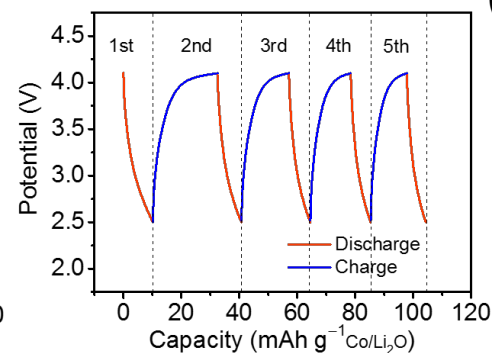
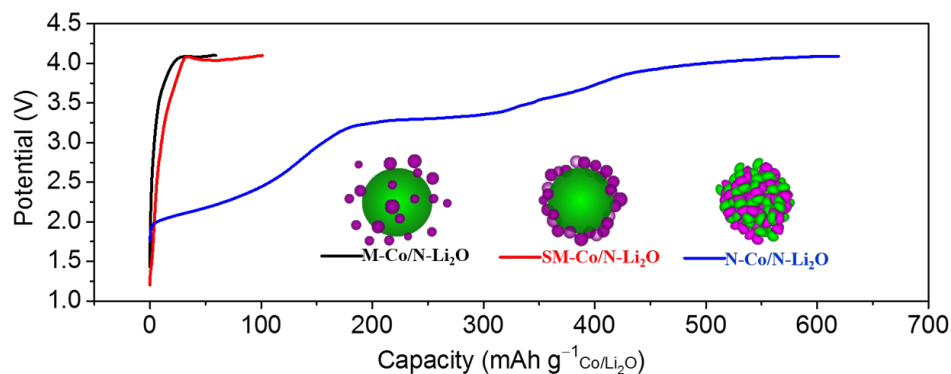
Cui group, *Nano Letters*  
16, 1497 (2016)

# Accomplishment

## $\text{Li}_2\text{O}$ /metal nanocomposite for cathode prelithiation

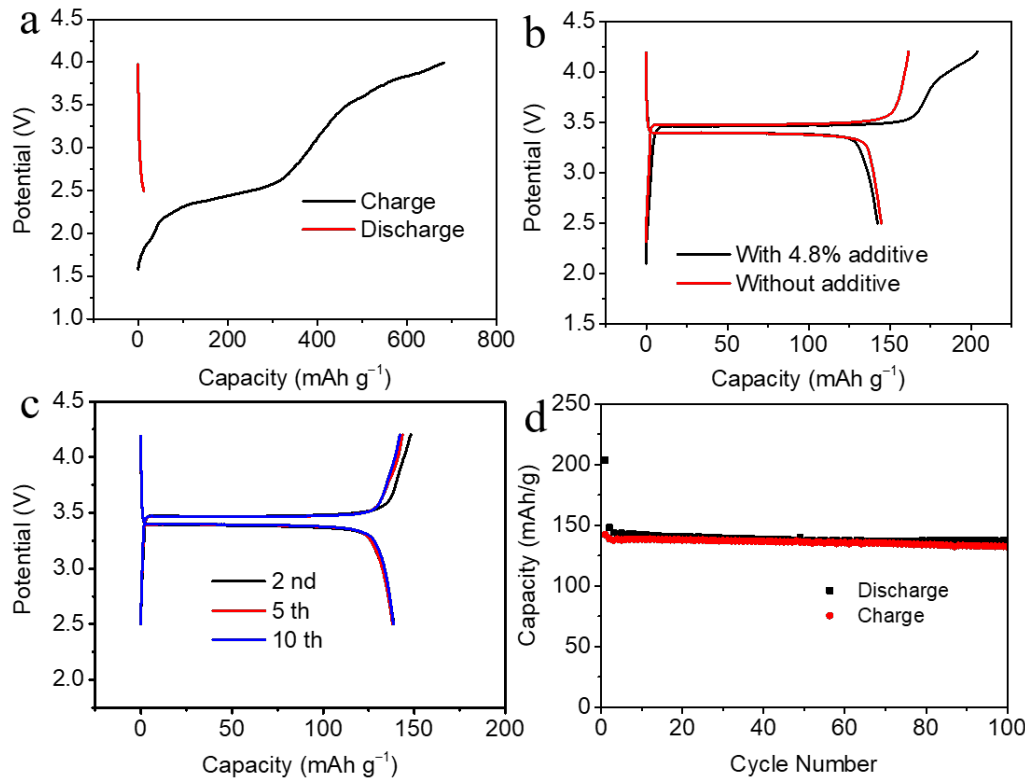
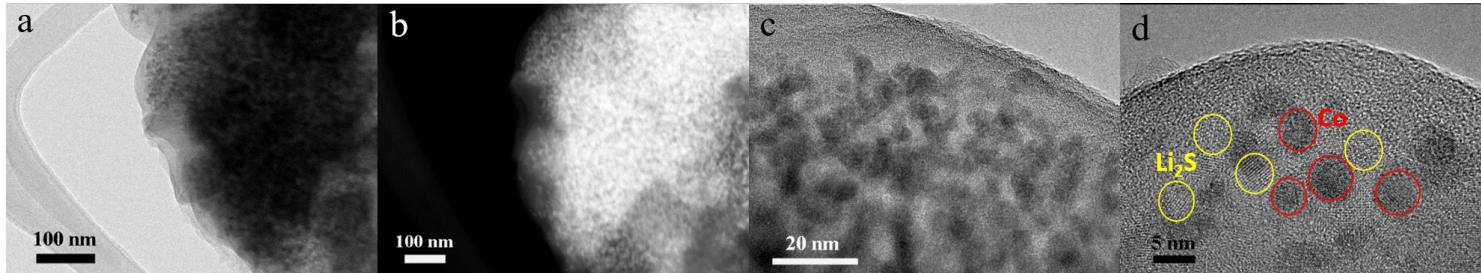


Cui group, *Nature Energy* 1, 15008 (2016)



# Accomplishment

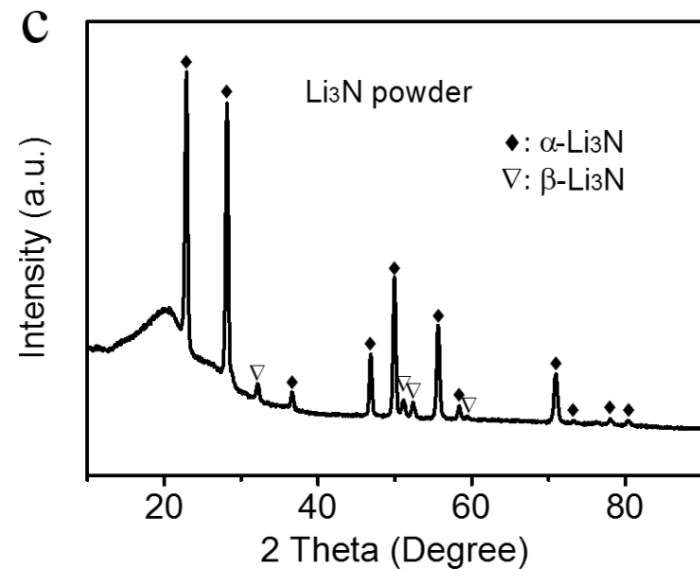
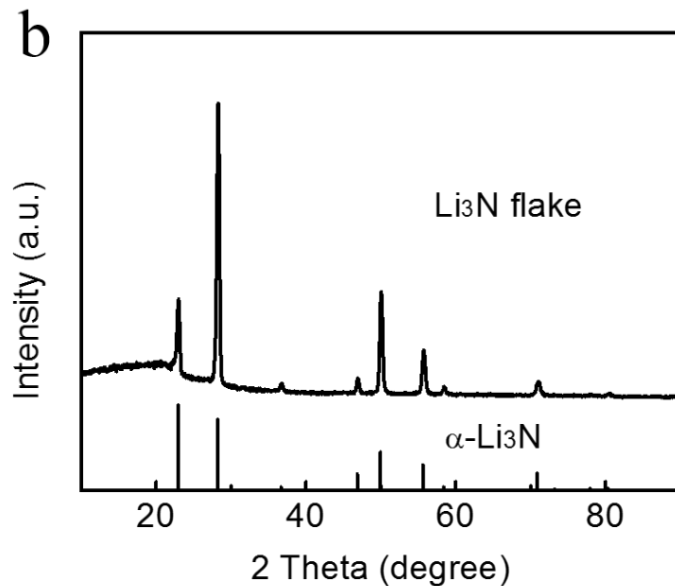
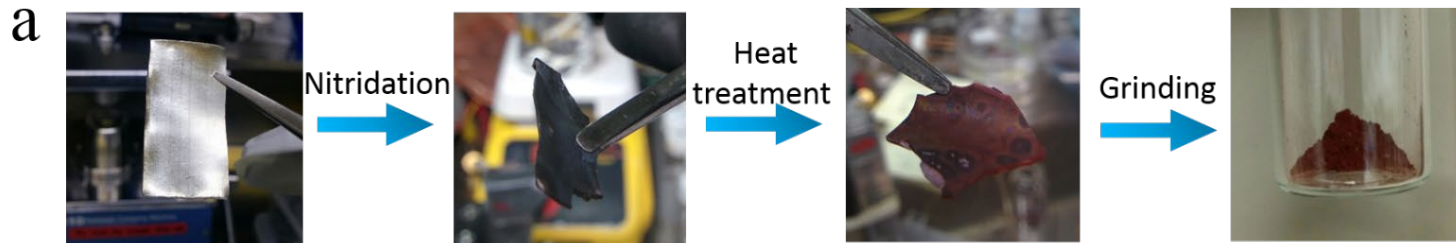
## Li<sub>2</sub>S/metal nanocomposite for cathode prelithiation



Cui group, *Advanced Energy Materials* 6, 1600154. (2016)

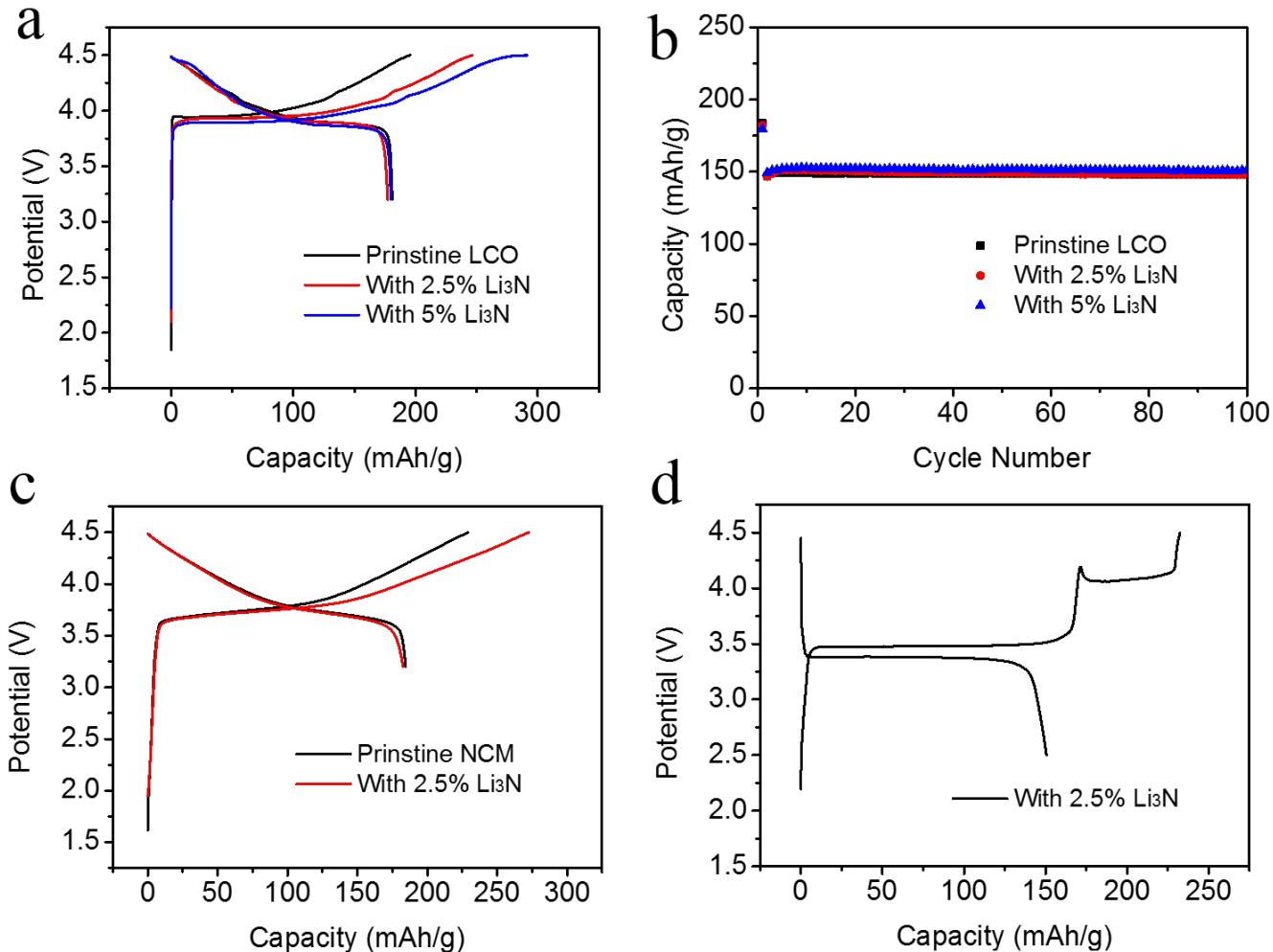
# Accomplishment

## Li<sub>3</sub>N nanocomposite for cathode prelithiation -Synthesis and characterizations



# Accomplishment

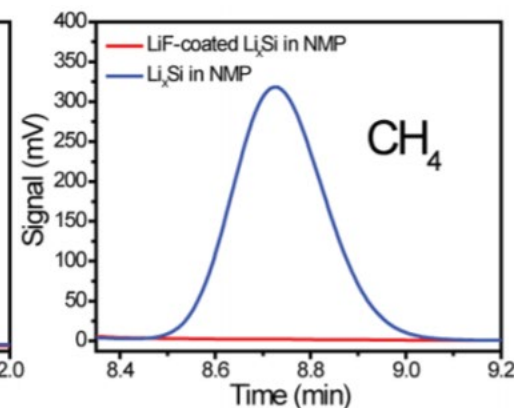
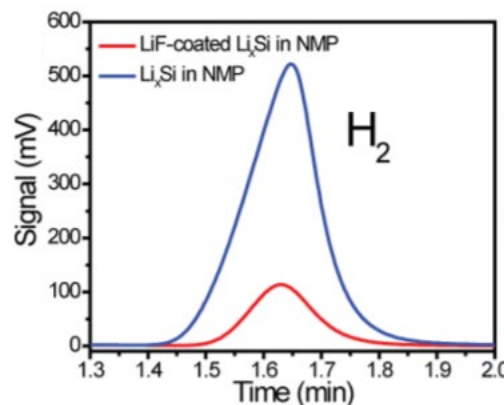
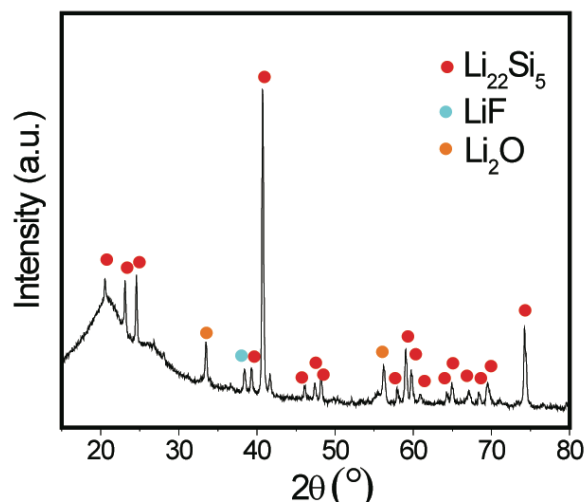
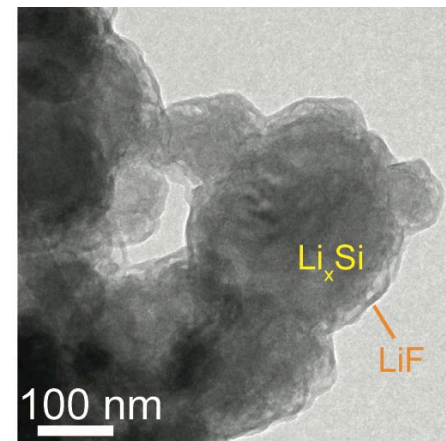
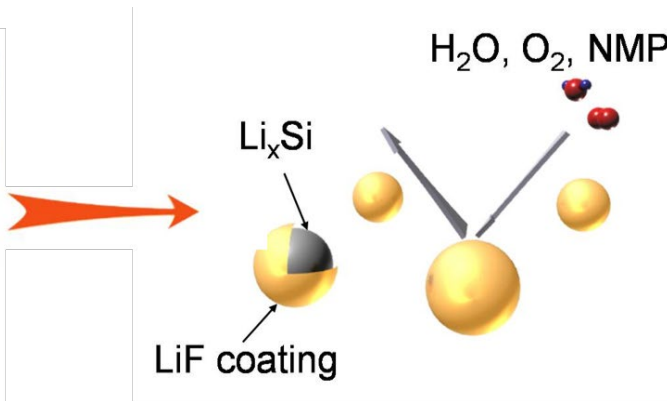
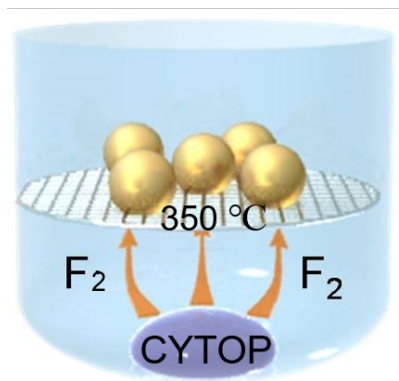
## Li<sub>3</sub>N nanocomposite for cathode prelithiation - Battery performance





# Accomplishment

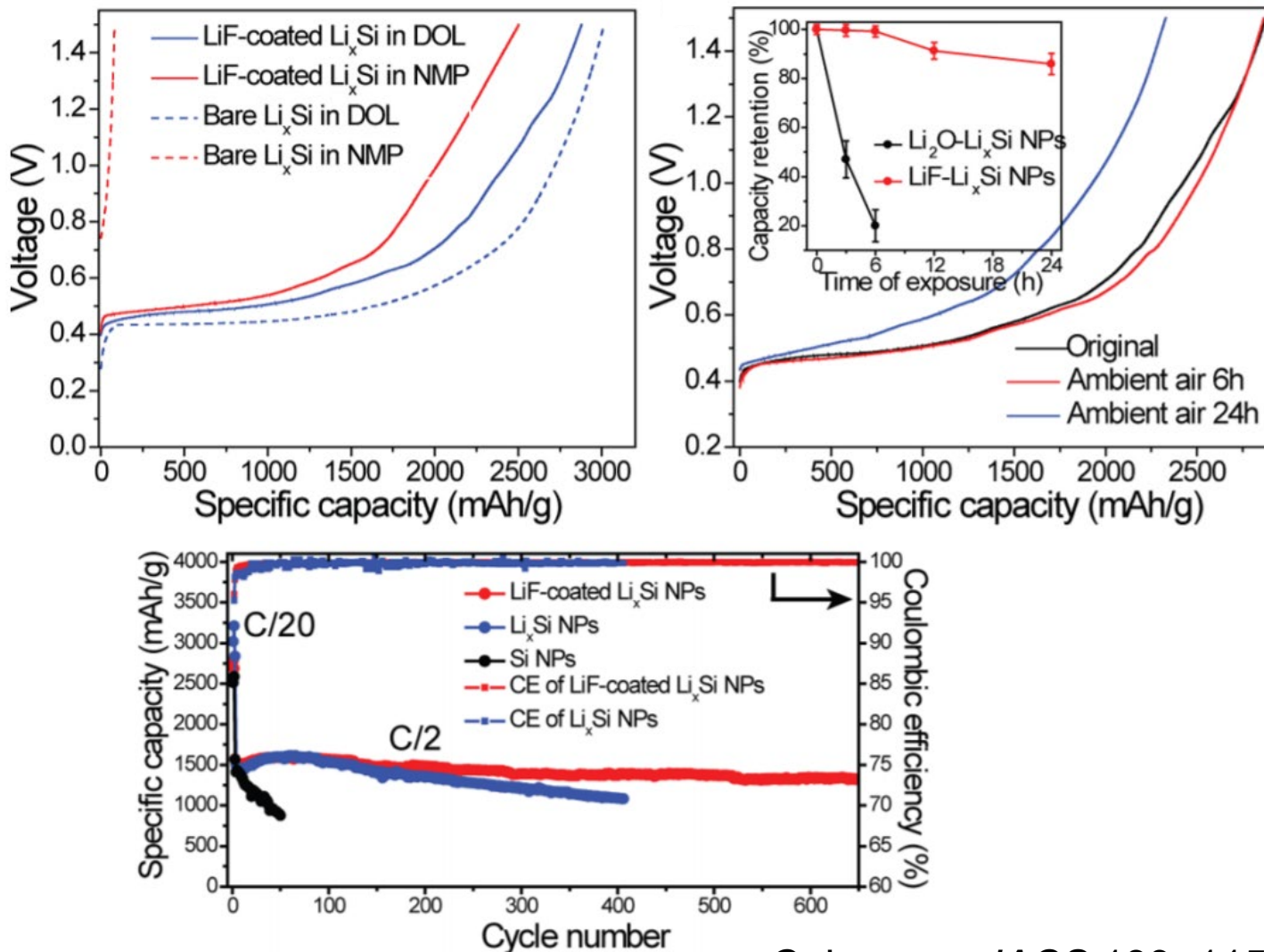
## NMP solvent-compatible $\text{Li}_x\text{Si}$ -LiF core-shell nanoparticles for anode prelithiation -Synthesis and characterizations



Reaction with NMP	$\text{H}_2$ (ppm)	$\text{CH}_4$ (ppm)
$\text{Li}_x\text{Si}$ without LiF	18145	938
$\text{Li}_x\text{Si}$ with LiF	2870	0

# Accomplishment

## NMP solvent-compatible $\text{Li}_x\text{Si}$ -LiF core-shell nanoparticles for anode prelithiation -Battery performance

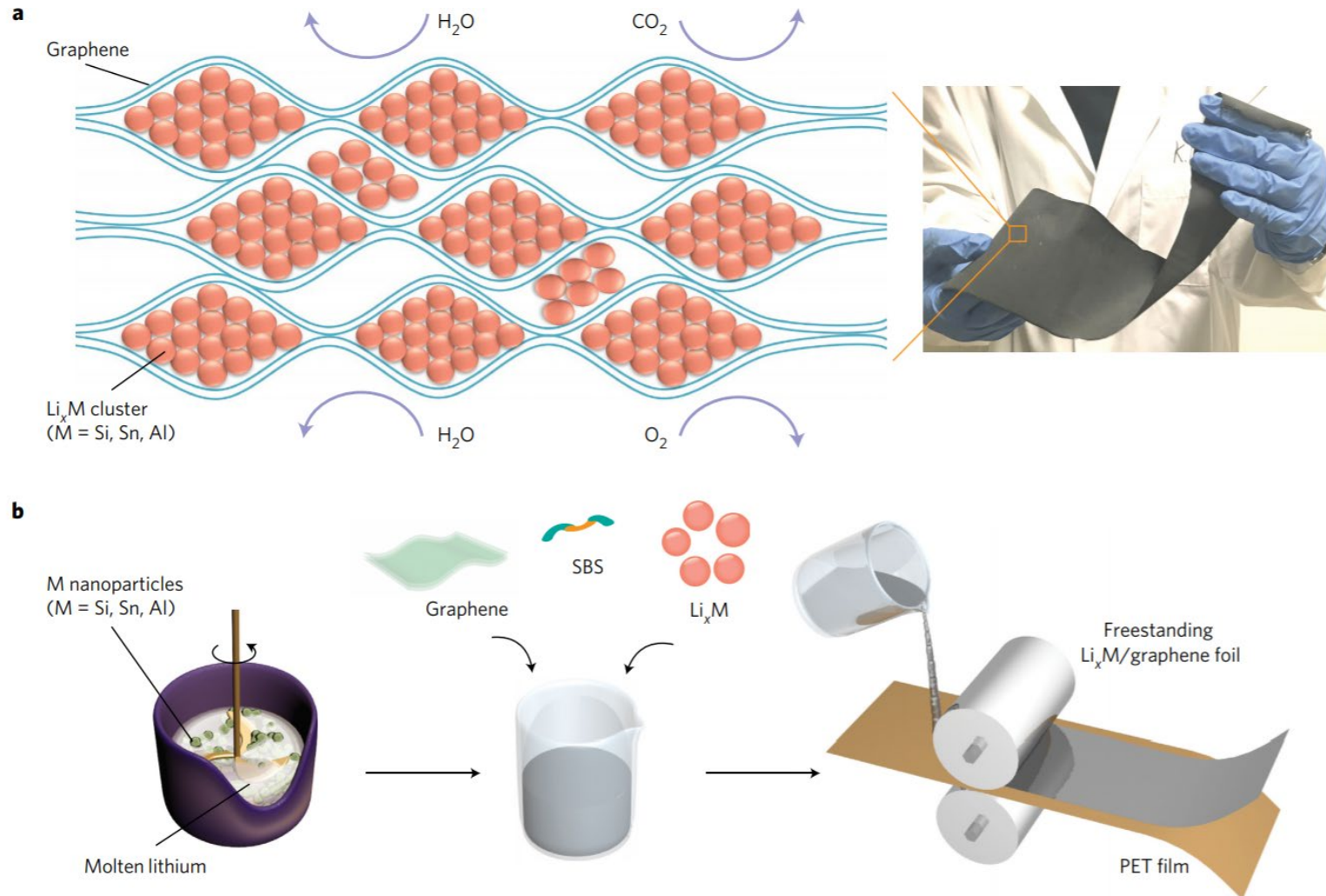


Cui group, *JACS* 139, 11550 (2017)



# Accomplishment

## Air-stable and freestanding lithium alloy/graphene foil for anode prelithiation -Synthesis and characterizations

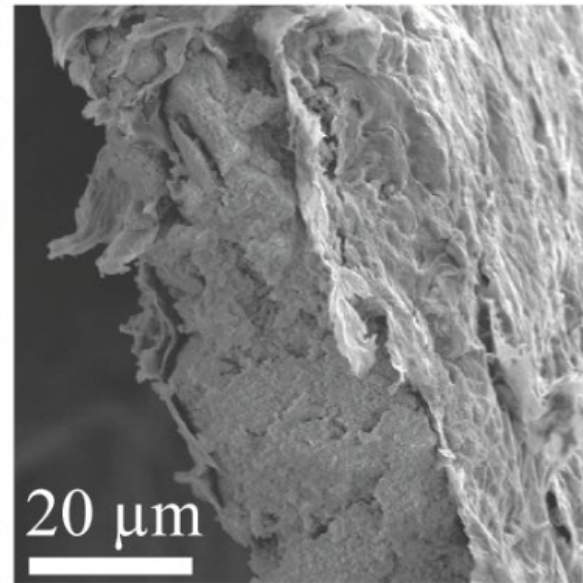
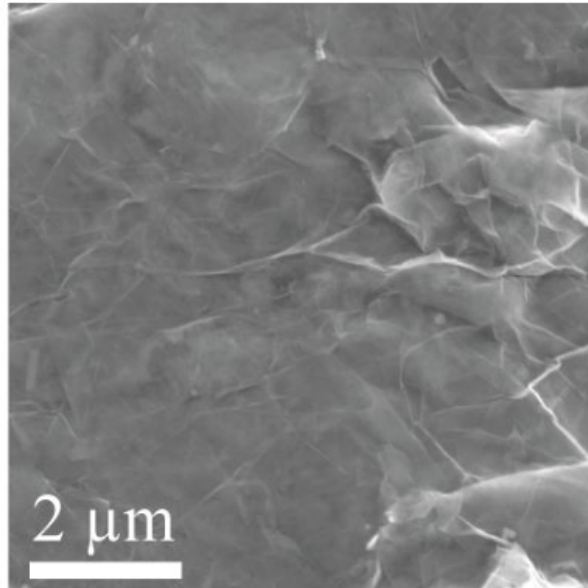
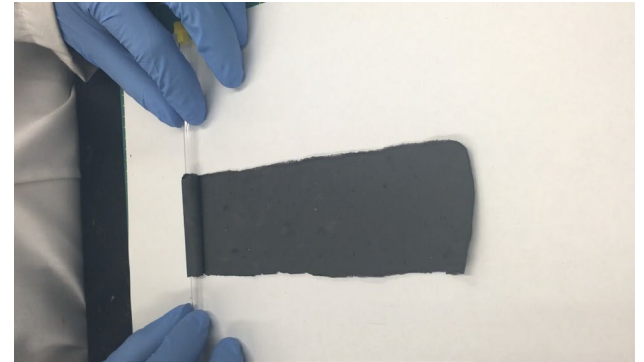
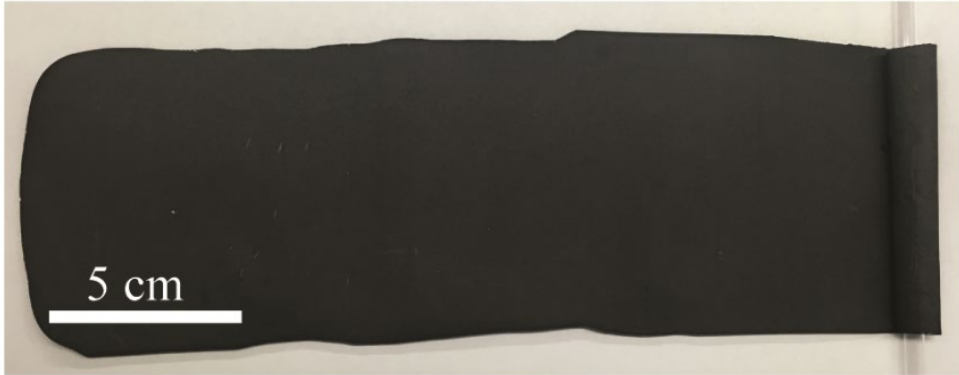


Cui group, *Nature Nanotechnology* 12, 993 (2017)

# Accomplishment

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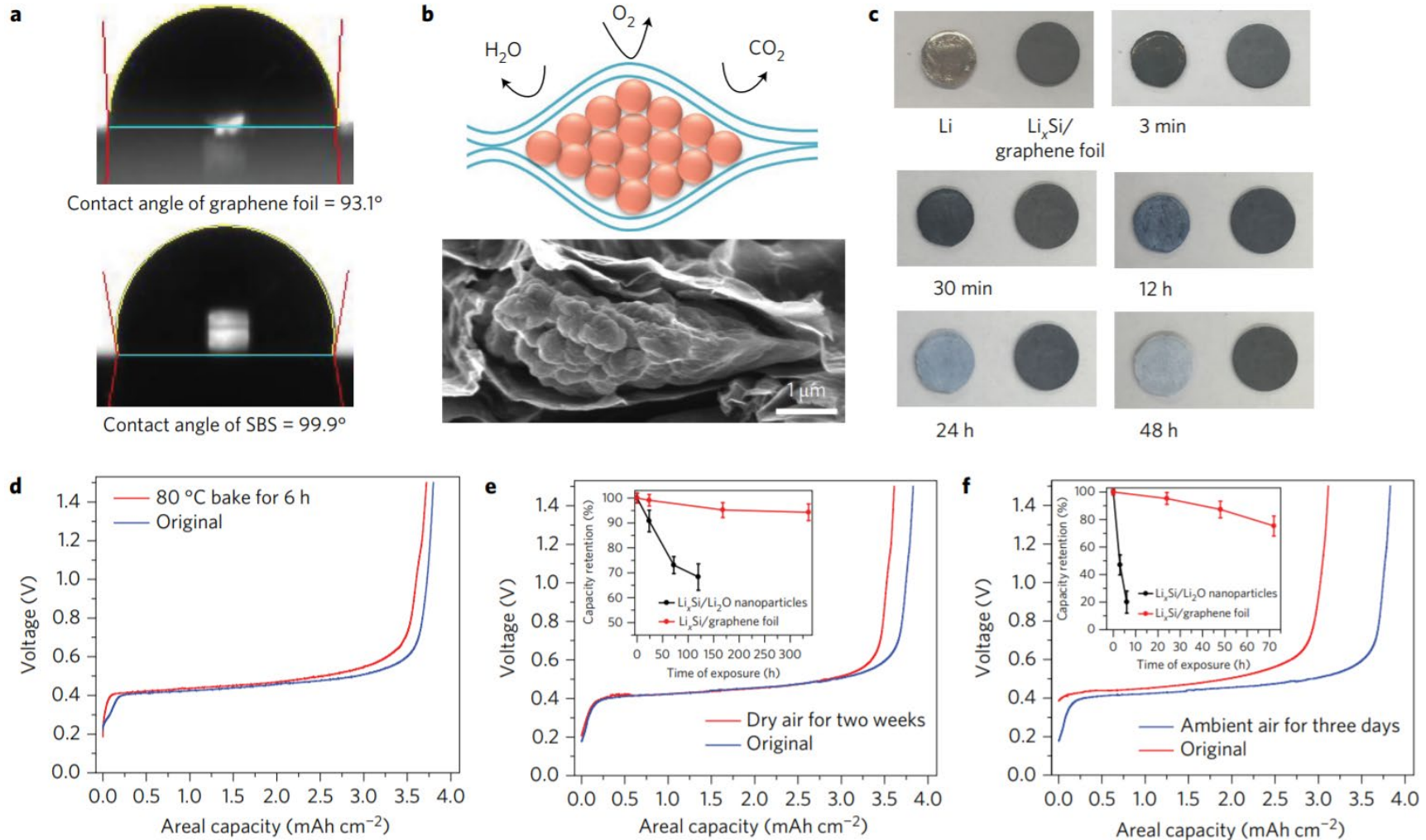
**Air-stable and freestanding lithium alloy/graphene foil for anode prelithiation**  
**-Synthesis and characterizations**



Cui group, *Nature Nanotechnology* 12, 993 (2017)

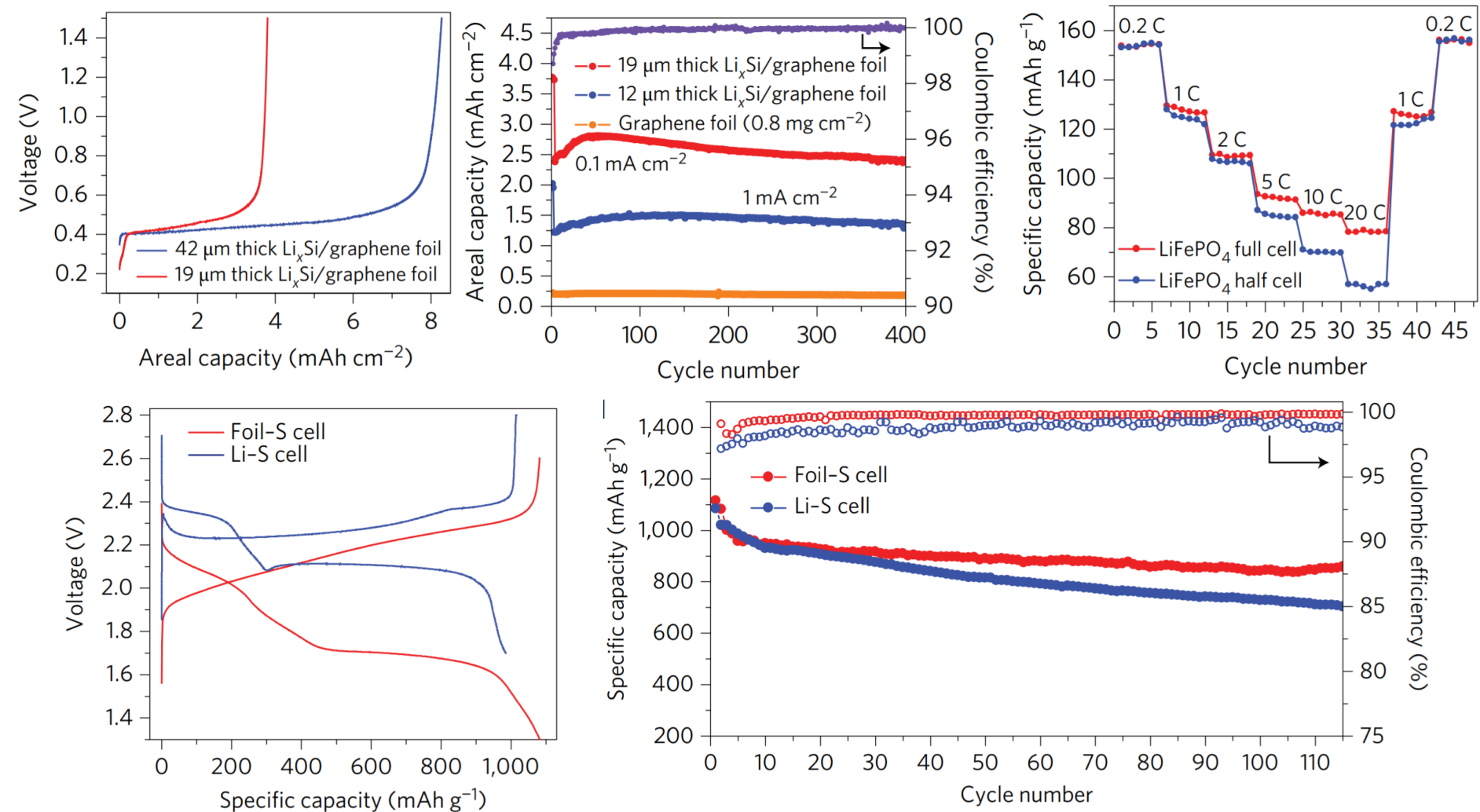
# Accomplishment

## Air-stable and freestanding lithium alloy/graphene foil for anode prelithiation -Stability in ambient air



# Accomplishment

## Air-stable and freestanding lithium alloy/graphene foil for anode prelithiation -Battery performance



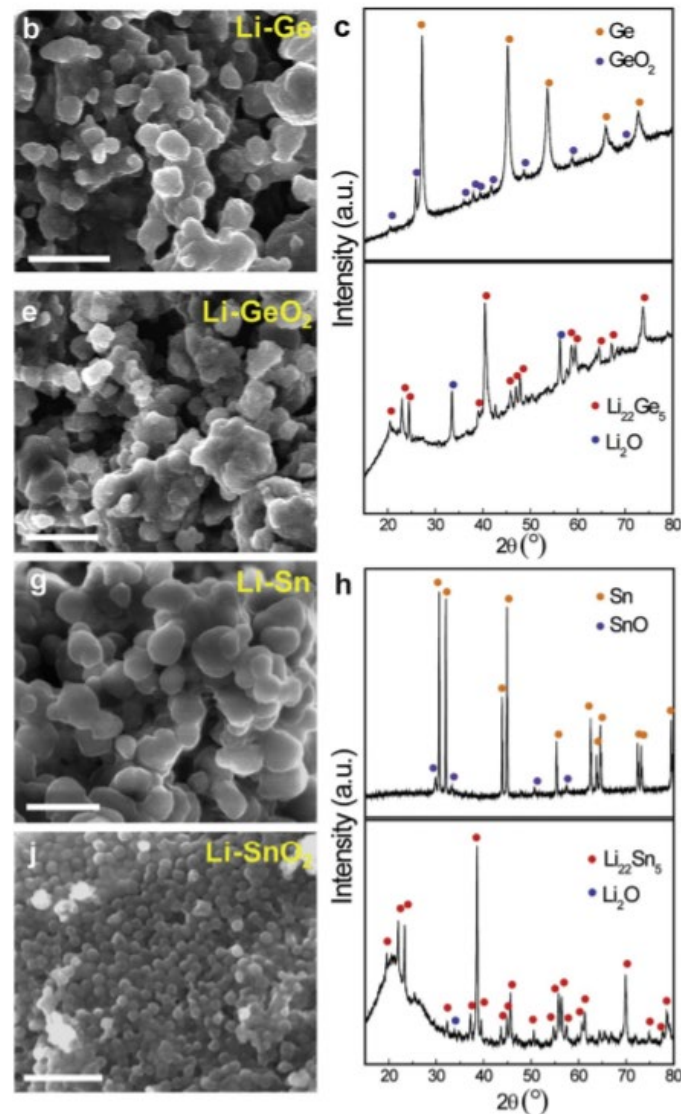
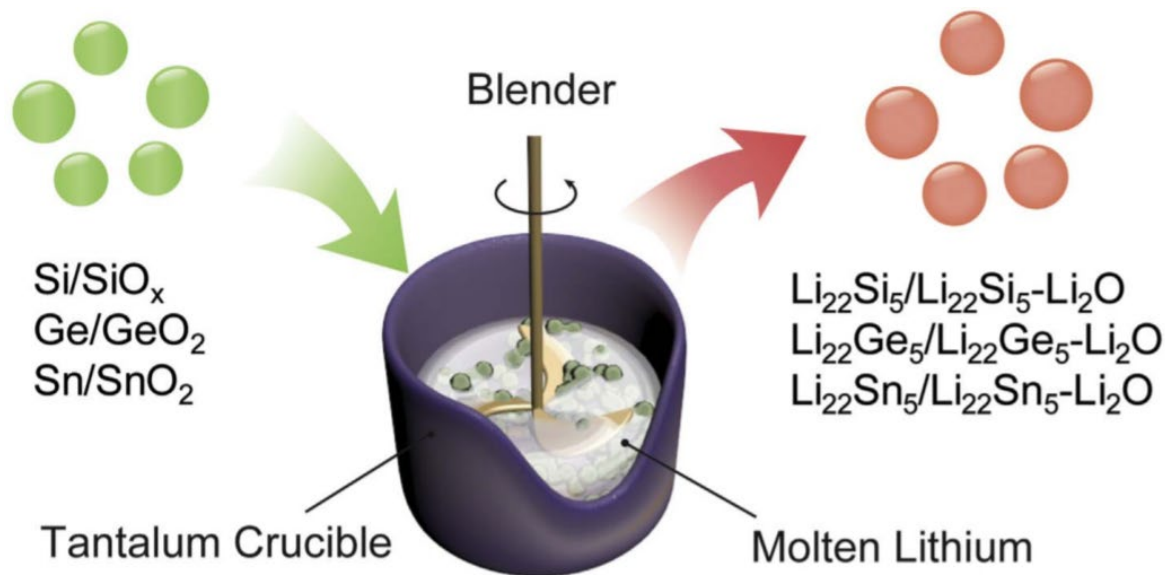
Cui group, *Nature Nanotechnology* 12, 993 (2017)



# Accomplishment

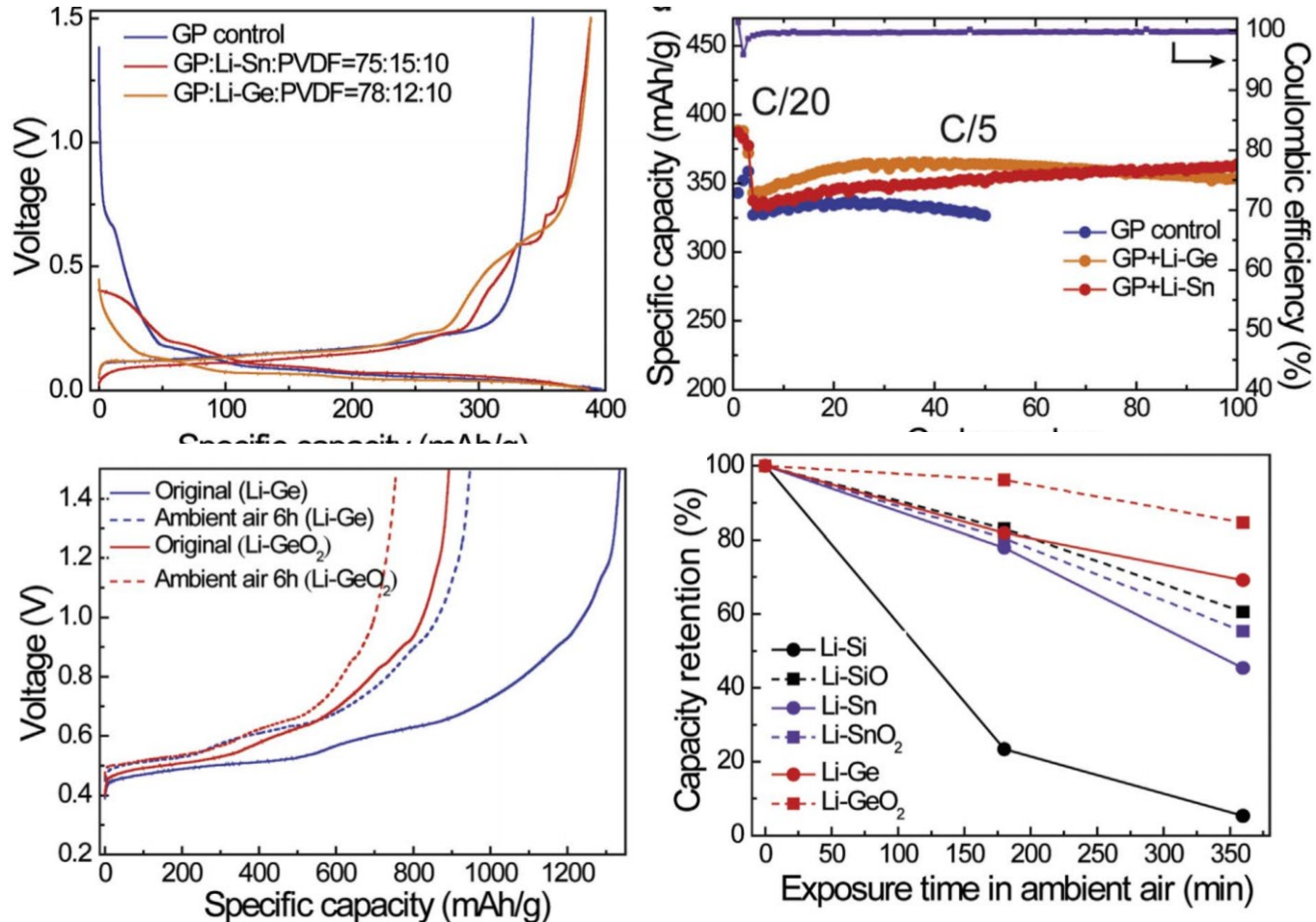
## General prelithiation approach for group IV elements and corresponding oxides

### -Synthesis and characterizations



# Accomplishment

## General prelithiation approach for group IV elements and corresponding oxides -Battery performance



# **Responses to Previous Year Reviewers' Comments**

Not applicable

# Collaboration and Coordination

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SLAC: In-situ X-ray, Prof. Mike Toney



Companies: Amprius Inc.

Stanford: Professor Zhenan Bao



# Remaining Challenges and Barriers

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- It is difficult to fabricate Li-rich anode materials with fine structures and stable cycling.
- It is difficult to realize cathode prelithiation process to be compatible with solvent processing.
- It is difficult to realize prelithiation process at the electrode level.

# Proposed Future Work

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- To improve the stability of prelithiation reagents in the slurry process by developing new solvent-binder combination.
- To improve the stability of prelithiation reagents in the ambient air condition by exploring different kinds of coatings and nanostructures.
- To explore other materials with high prelithiation capacity and stability.
- To understand the interaction between molecule in the air and different coatings of the prelithiation reagents.
- To synthesize Li-rich anode materials with fine structures and stable cycling and then pair them with high capacity Li-free cathode materials.
- To explore the prelithiation process at the electrode level instead of particle level through a facile method.

# Summary

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- **Objective and Relevance:** The goal of this project is to increase 1<sup>st</sup> cycle Coulombic efficiency of lithium ion batteries via both anode and cathode prelithiation.
- **Approach/Strategy:** This project combines advanced materials synthesis, characterization, battery assembly and testing, which has been demonstrated to be highly effective.
- **Technical Accomplishments and Progress:** This project has produced many significant results, meeting milestones. They include identifying the key issues in prelithiation, using rational materials design, synthesizing and testing, and developing scalable and low-cost methods. The results have been published in top peer-reviewed scientific journals. The PI has received numerous invitations to speak in national and international conferences.
- **Collaborations and Coordination:** The PI has established a number of highly effective collaborations.
- **Proposed Future Work:** Rational and exciting future has been planned.

59. (Invited) “Nanoscale Design and Cryogenic Electron Microscopy for Energy Storage”, American Chemical Society Spring Meeting, New Orleans, March 18-22, 2018.
60. (Invited) “Emerging Materials Selection and Design for Batteries with High Energy Density, Ultralong Cycle Life and Excellent Safety”, International Battery Seminar, Fort Lauderdale, March 26-29, 2018.
61. (Invited) “Materials design of Li-metal host and interface and characterization with cryo electron microscopy”, US-Germany Energy Storage Workshop, Washington DC, March 26-27, 2018.
62. (Invited) “Nanoscale Design and Cryogenic Electron Microscopy for of Energy Storage”, Materials Research Society Spring Meeting, Symposium EN14, Phoenix, Arizona, Apr 2-6, 2018.
63. (Invited) Yi Cui “New Battery Technology for Electrical Vehicles and Grid Scale Energy Storage” the 2018 US-China Green Energy Summit (2018UCGES), Aug 3-4, 2018, Hyatt Regency, San Francisco Airport.
64. (Invited) “How Far Can Batteries Go” Global Energy Forum, Stanford University, Nov. 1-2, 2018.
65. (Invited) “Nanoscale Design for Lithium-Sulfur Batteries”, Materials Research Society Fall Meeting, Symposium ET09, Boston, Nov. 25-30, 2018.
66. (Invited) “Nanoscale Composite Polymer Electrolyte Batteries”, Symposium ET01 Materials Research Society Fall Meeting, Boston, Nov. 25-30, 2018.
67. (Invited) “Nanotechnology for Energy, Environment and Textile”, Seminar, Peking University School of Materials Science at Shenzhen, China, Jan 14, 2018.
68. (Invited) “Nanotechnology for Energy, Environment and Textile”, seminar, Dalian Institute of Chemical Physics, Dalian, China, Jan 16, 2018.
69. (Invited) “Nanotechnology for Energy, Environment and Textile”, seminar, Dalian Institute of Technology, Dalian, China, Jan 19, 2018.
70. (Invited) “Nanotechnology for Energy, Environment and Textile”, seminar, UC Irvine, Feb. 2, 2018.
71. (Invited) “Pathways of Batteries Towards Sustainable Electric Transportation and Stationary Storage”, Stanford Energy Seminar, Stanford University, Feb. 12, 2018.
72. (Invited) “Nanotechnology for Energy, Environment and Textile”, seminar, Tsinghua University, Beijing, China, Mar 5, 2018.
73. (Invited) “Nanotechnology for Energy, Environment and Textile”, seminar, Beijing Institute of Technology, Beijing, China, Mar 5, 2018.
74. (Invited) “Nanotechnology for Energy, Environment and Textile”, seminar, China National Center of Nanotechnology, Beijing, China, Mar 6, 2018.
75. (Invited) “Nanotechnology for Energy, Environment and Textile”, seminar, Institute of Processing, Chinese Academy of Sciences, Beijing, China, Mar 6, 2018.
76. (Invited) “Nanotechnology for Energy, Environment and Textile”, seminar, University of Science and Technology of Beijing, Beijing, China, Mar 7, 2018.
77. (Invited) “Nanotechnology for Energy, Environment and Textile”, seminar, Research Institute of China Petroleum, Beijing, China, Mar 7, 2018.
78. (Invited) “Nanotechnology for Energy, Environment and Textile”, seminar, Institute of Physics and Chemistry, Chinese Academy of Sciences, Beijing, China, Mar 8, 2018.
79. (Xinda Lectureship) “Nanotechnology for Energy, Environment and Textile”, Peking University, Beijing, China, Mar 9, 2018.
80. (Invited) “Materials Design for High Energy Batteries towards Electric Transportation”, seminar, General Motors, Warren, Michigan, March 16, 2018.
81. (Invited) “Nanomaterials Design for Energy Storage and Catalysis” Northwestern University, Materials Science Seminar, Oct. 2, 2018.
82. (Invited) "Batteries Now and Future". The Stanford Chinese Faculty & Family Club (SCFFC), Quarterly Speaker Series, Nov. 4, 2018.